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Analysis of Heterocatalytic Reactor Bed Based on Catalytic Pellet Models

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Abstract: Nowadays models are applied in almost every field of our life to predict how real systems behave. Obviously, models are used also in process industry, e.g. to improve the safety of chemical plants. Models can be applied to design a new process, to reveal hidden capacities of the process, to support process operator in avoiding any undesired situations, etc. However, the design and operation of such reactors are usually based on empirical criteria. Complex models can be implemented and solved in commercial Computational Fluid Dynamics (CFD) software (e.g. COMSOL Multiphysics). CFD has become a standard tool for analyzing various situations where fluid flow has a significant effect on the studied processes. The goal of this paper is to model a catalytic bed based on the model of individual catalyst pellets. The proposed model take into account with the mass, heat and momentum transport as well as the reactions in a catalytic pellet of an industrial fixed-bed tube reactor. Different ways for aggregating pellet models were compared and evaluated with COMSOL Multiphysics. The catalytic pellet model was exported using MATLAB-COMSOL connection, and the catalytic bed model was implemented and tested using MATLAB. The CFD applications in general need more and more advanced hardware (e.g. more random access memory - RAM) to solve the developed models, however our approach requires significantly less hardware resources.

Keywords – *CFD, heterocatalytic system, pellet model, reactor bed model,*

I. INTRODUCTION

Nowadays development of flow field in process equipments is a key issue in design of process units. With a well designed flow field the overall process performance and the safety of operation can be increased at the same time. To support this step, different modeling software packages and flow sheet simulators have been developed in last decades. As computational capacity of

computers has being increased the role of computational fluid dynamics (CFD) in applied modeling technics and method has significantly grown since the first computer was switched on [1].

CFD can be considered as a collection of numerical solvers for partial differential equations. With applying CFD codes more complicated problems in more complex structures can be investigated in details than with other modeling technics. In recent years many papers applying CFD technique to analyze some phenomena or to design a reactor have been published. The most frequently investigated phenomena are the pressure drop, flow profile [2,3], heat transfer and mass transfer [4,5,6] for a reactor, the same for a catalytic pellet [8] and intra-particle diffusion effects [9] too. These examples show that CFD is capable to simulate transport processes and it can be applied as a part of complex mathematical models.

Our purpose was to study heterocatalytic systems. The model development process will be introduced in case of an industrial heterocatalytic reactor with highly exothermic reactor. COMSOL Multiphysics can give an effective support since it can be applied to solve models which consist higher order partial differential equations with finite elements method. In this article COMSOL Multiphysics is applied to calculate the development of state variables in a packed bed tube reactor.

II. CASE STUDY

The simplest developed reactor models were the one-dimensional quasi-single phase steady-state and dynamical models. It was followed by a two-phase dynamical model since heterocatalytic processes can be more adequately described by applying separate solid and gas phases.

These models can be applied in the analysis of any possible situation in the system, such as reactor runaway. In 1D models the state-variables are calculated along the catalyst bed. However the radial diffusion cannot be considered without expanding dimensions. 2D models take into consideration the geometry and the evolved flow field in the system. Since the geometry of the reactor is symmetrical some simplifications can be applied to decrease the necessary calculation cost during the solution of models. The analysis shows the significant heat sources (hot spots) and the unusual flow fields (back mixing, turbulent waves). These types of results can be suitable in optimization of reactor efficiency [5,6]. The most complex developed model is the 2D, two-phase model. In this model both solid and gas phases are considered. The detailed mathematical description of the pellet has been performed in COMSOL Multiphysics. It can be applied to investigate the influence of diffusion and the velocity field on the reactor temperature and the rate of transport processes between the considered phases. Based on the modelling possibilities the pellet model can be applied to design the shape and size distribution of catalyst pellets, to get a closer look to specific locations of the catalyst bed, to analyze the possible development of runaway in a single catalyst pellet, and to build a catalytic bed from them. Unfortunately, for solving of this type of complex reactor models requires horrible hardware capacity.

III. PELLET MODEL

In our previous works, a quasi-single phase mathematical description of the reactor, and 2D models, which take into consideration the geometry and the evolved flow field in the system were

introduced [4,7,10]. The investigated, vertically positioned reactor contains a large number of tubes filled with catalyst. Our concept is the integration of a catalyst pellet model, implemented in COMSOL Multiphysics into the catalytic bed using a pellet by pellet building technique developed in MATLAB. The first step to accomplish our objective is the implementation of pellet model in the applied CFD code. All component mass balances in the gas phase contain terms to calculate convective and conductive transport processes and naturally the mass transport process between the solid and gas phase:

$$\frac{d(B^G \cdot c_i^G)}{dx} + \frac{d(B^G \cdot c_i^G)}{dy} = -D_i^G \frac{d^2 c_i^G}{dx^2} +$$

$$-D_i^G \frac{d^2 c_i^G}{dy^2} - A^{GS} \cdot \beta_i^{GS} \cdot (c_i^S - c_i^G) \quad (1)$$

where B^G – flow rate of gas phase; $i = \{A; B; C\}$; c_i^G – concentration of i^{th} -component in gas phase; x – reactor length; D_i^G – diffusion coefficient of i^{th} -component in gas phase; A^{GS} – interface area between the gas and solid phase; β_i^{GS} – mass transfer coefficient between the gas and solid phase of i^{th} -component; c_i^S – concentration of i^{th} -component in solid phase. Of course the convective term in solid phase is missing:

$$D_i^S \frac{d^2 c_i^S}{dx^2} + D_i^S \frac{d^2 c_i^S}{dy^2} = -A^{GS} \cdot \beta_i^{GS} \cdot$$

$$\cdot (c_i^S - c_i^G) + V^S \cdot v_i \cdot r \quad (2)$$

where D_i^S – diffusion coefficient of i^{th} -component in solid phase; V^S – the volume of solid phase; v_i – the stoichiometric coefficient of i^{th} -component; r – reaction rate which is calculated by the following correlation:

$$r = k_0 \cdot e^{-\frac{E_A}{RT^S}} \quad (3)$$

where k_0 – preexponential factor; E_A – activation energy; R – ideal gas constant; T^S – temperature of solid phase. Terms in both equations are the same as those considered in component balances:

$$\rho^G \cdot c_p^G \cdot B^G \cdot \left(\frac{dT^G}{dx} + \frac{dT^G}{dy} \right) = A^{GS} \cdot \alpha^{GS} \cdot (T^S - T^G) - \lambda^G \cdot \left(\frac{d^2T^G}{dx^2} + \frac{d^2T^G}{dy^2} \right) \quad (4)$$

where ρ^G – density of gas phase; c_p^G – heat capacity of gas phase; T^G – temperature of gas phase; λ^G – heat conduction coefficient in gas phase; α^{GS} – heat transfer coefficient between the gas and solid phase. And finally the heat balance of solid phase:

$$\lambda^S \cdot \left(\frac{d^2T^S}{dx^2} + \frac{d^2T^S}{dy^2} \right) = -A^{GS} \cdot \alpha^{GS} \cdot (T^S - T^G) + V^S \cdot r \cdot (-\Delta H_r) \quad (5)$$

where λ^S – heat conduction coefficient in solid phase; ΔH_r – reaction heat. This is just a short description of model equations to show the structure of pellet model. All physical properties are calculated as the function of temperature. These equations have been applied to implement a pellet model in COMSOL.

IV. CATALYTIC BED MODEL

In the following the pellet model implemented in COMSOL was used to build the model of the catalyst bed. Using the high-level connection between MATLAB and COMSOL, the implemented pellet model can be exported to MATLAB as a function. The user interface of MATLAB let you to modify the pellet model, and to build up a structure from them. As Figure 1 shown, the first pellet outlet properties are boundary averages, and these results became the next pellet inlet parameters. Finally, a linear catalytic bed was built and the first investigations were conducted on it.

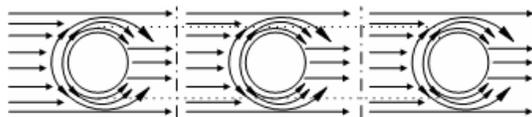
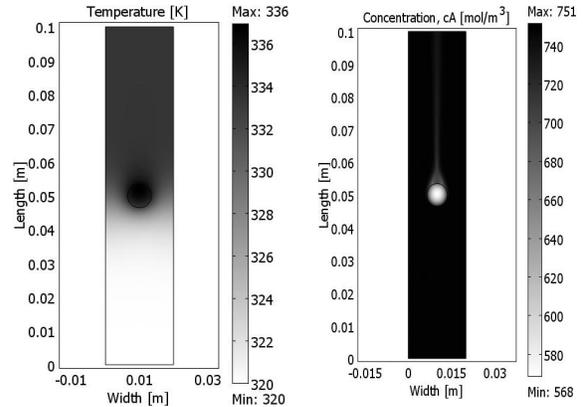


Figure 1. The visualization of the implemented pellet structure

V. RESULTS AND DISCUSSIONS

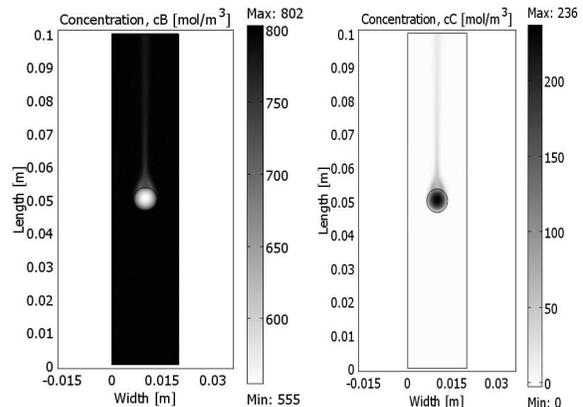
For the COMSOL Multiphysics pellet model the value of state variables in front of the reactor were given as boundary conditions in the solution of the pellet model. Obtained results of the pellet model are plotted in Figure 2.

In Figure 2.a the temperature surface can be seen and as it was expected the highly exothermic reaction taking place in catalyst increases the temperature of the pellet and it warms up the flowing gas.



a) Temperature in the catalyst pellet and in its surroundings

b) Concentration of A component in the catalyst pellet and in its surroundings



c) Concentration of B component in the catalyst pellet and in its surroundings

d) Concentration of C component in the catalyst pellet and in its surroundings

Figure 2. State-variables along the investigated part of reactor in steady-state

Due to the diffusion in the catalyst pellet the concentration of reagents decrease in the direction to the centre of the pellet (see Figure 2.b and Figure 2.c); while the concentration of the product changes in the opposite direction as shown in Figure 2.d.

In this model all components can be connected to the same active sites on the catalyst, however in practise there is a competition between components and the rate and the equilibrium of the adsorption determine the concentration in the catalyst. Since only the temperature profile and properties of inlet and outlet flows are measured during the reactor operation, the catalyst pellet model cannot be validated without being integrated into the reactor simulator.

As it can be seen in Figure 2.d the outlet concentration of the product is very low. This suggests that the parameters applied to calculate the rate of adsorption must be modified. Still the characteristic of changes in case of all the calculated state variables are the same as we expected.

This pellet model have been exported to MATLAB, and used as a function. The MATLAB software ensures wide freedom for its users, and with its built-in function it makes possible to implement a catalytic bed pellet by pellet. The effect of the catalyst volume distribution on the conversion was analysed. The number of pellets was increased at near constant catalytic volume. Experiments presented in literature [11] suggest the smaller the pellets are, the better the conversion/volume index is. Figure 3 and 4 show the reagent and Figure 5 the product concentration along the bed. As the number of particles is increasing higher and higher conversion can be achieved. Due to the exothermic reactions the temperature will be higher in the outlet than in the inlet. Figure 6 show that the temperature has a nearly linear relationship with the conversion.

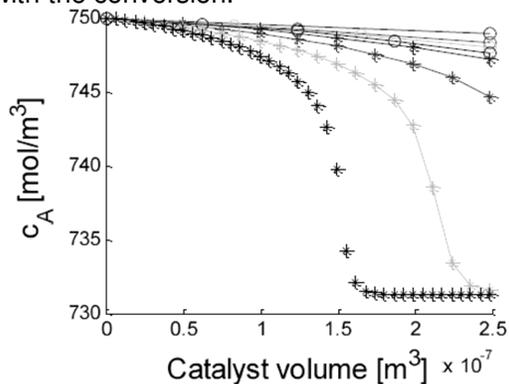


Figure 3. The concentration of A component in the catalytic bed

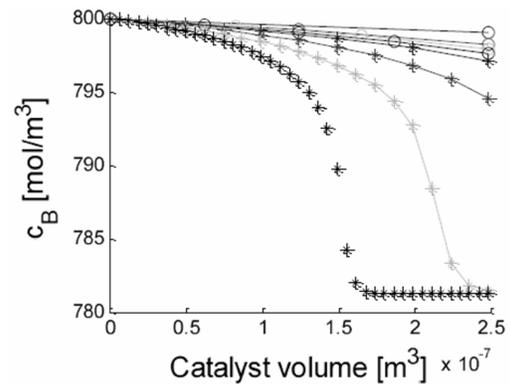


Figure 4. Concentration of B component in the catalytic bed

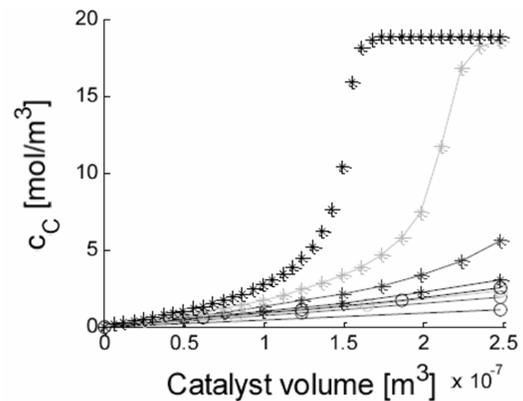


Figure 5. Concentration of C component in the catalytic bed

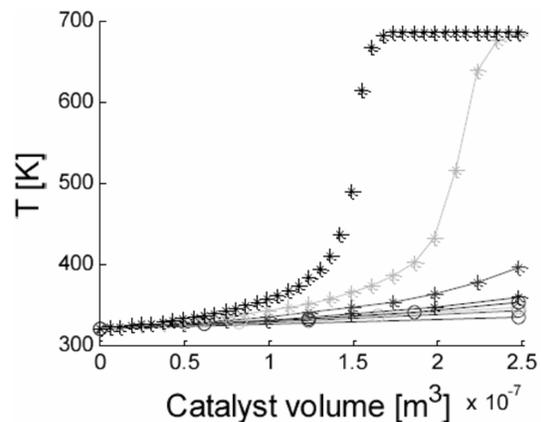


Figure 6. Temperature in the catalytic bed

VI. CONCLUSIONS AND FUTURE WORK

To monitor an operation of a tube reactor the process operators usually do not have much information, just some

temperature measurements along the reactor and composition of inlet and outlet flows. Hence, a reactor simulator can be very useful to follow the operation and to help process operators, e.g. to avoid development of reactor runaways or to optimize reactor operation. The two dimensional model of catalyst pellet with its close surroundings was implemented into COMSOL Multiphysics. This pellet model can be utilised to design the shape and size distribution of catalyst pellets, to get a closer look to specific locations of the catalyst bed, and to build a catalytic bed from them. The catalytic bed was implemented pellet by pellet and the result of the first tests fulfilled the expectations. Our vision is a flexible, pellet by pellet catalytic bed model, which can be used for the design, study and development of catalytic pellets and beds with reduced calculation cost.

VII. ACKNOWLEDGEMENT

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VIII. REFERENCES

- [1] Ranade V. V.: Computational flow modelling for chemical reactor engineering, Academic Press (2002)
- [2] A. Jafari , P. Zamankhan, S.M. Mousavi, K. Pietarinen, Modeling and CFD simulation of flow behavior and dispersivity through randomly packed bed reactors, Chemical Engineering Journal 144 (2008) 476–482
- [3] H. P. A. Calis, J. Nijenhuis, B. C. Paikert, F. M. Dautzenberg, C. M. van den Bleek, CFD modelling and experimental validation of pressure drop and flow profile in a novel structured catalytic reactor packing, Chemical Engineering Science 56 (2001) 1713-1720
- [4] Varga, T., Abonyi, J., Chován, T., Nagy, L., Szeifert, F., Réti, J., Proceedings Műszaki Kémiai Napok '06, , 45-48 (2006) (in Hungary)
- [5] Tarek M. Moustafa, Mohamed Abou-Elreesh and Seif-Eddeen K. Fateen, Modeling, Simulation, and Optimization of the Catalytic Reactor for Methanol Oxidative Dehydrogenation COMSOL Conference 2007, Boston (2007)
- [6] Maryam Ghadrhan*,1, Hamid Mehdizadeh Mathematical Investigation and CFD Simulation of Monolith Reactors: Catalytic Combustion of Methane, COMSOL Conference Hannover (2008)
- [7] Gy. Rádi, T. Varga, T. Chován., Műszaki Kémiai Napok '09, Proceedings, 234-238 (2009) (in Hungary)
- [8] Anuradha Nagaraj¹ and Patrick L. Mills², Analysis of Heat, Mass Transport, and Momentum Transport Effects in Complex Catalyst Shapes for Gas-Phase Heterogeneous Reactions Using COMSOL Multiphysics, COMSOL Conference Boston (2008)
- [9] A. Guardo, M. Casanovas, E. Ramírez, F. Recasens, I. Magaña, D. Martínez, M.A. Larrayoz, CFD modeling on external mass transfer and intraparticle diffusional effects on the supercritical hydrogenation of sunflower oil, Chemical Engineering Science 62 (2007) 5054 – 5061
- [10] Varga T., Szeifert F., Chován T., Réti J., Műszaki Kémiai Napok '07, Proceedings, (2007) 113-117 (in Hungary)
- [11] J.M. Berty, Experiments in Catalytic Reaction Engineering, Studies in Surface Science and Catalysis, 124, Elsevier, 1999